ORGANIC LETTERS

2013 Vol. 15, No. 18 4678–4681

Aogacillins A and B Produced by *Simplicillium* sp. FKI-5985: New Circumventors of Arbekacin Resistance in MRSA

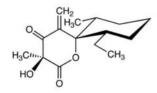
Keiko Takata,^{†,⊥} Masato Iwatsuki,^{‡,⊥} Tsuyoshi Yamamoto,[‡] Tatsuya Shirahata,[§] Kenichi Nonaka,[‡] Rokuro Masuma,[‡] Yoichi Hayakawa,[†] Hideaki Hanaki,[‡] Yoshinori Kobayashi,[§] George A. Petersson,[□] Satoshi Ōmura,^{*,‡} and Kazuro Shiomi*,[‡]

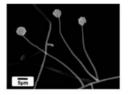
Faculty of Pharmaceutical Sciences, Tokyo University of Science, 2641 Yamazaki, Noda, Chiba, 278-8510, Japan, Kitasato Institute for Life Sciences, Kitasato University, 5-9-1 Shirokane, Minato-ku, Tokyo 108-8641, Japan, School of Pharmacy, Kitasato University, 5-9-1 Shirokane, Minato-ku, Tokyo 108-8641, Japan, and Hall-Atwater Laboratories of Chemistry, Wesleyan University, Middletown, Connecticut 06459-0180, United States

shiomi@lisci.kitasato-u.ac.jp; omuras@insti.kitasato-u.ac.jp

Received July 13, 2013; Revised Manuscript Received August 23, 2013

ABSTRACT





Aogacillin A

Simplicillium sp. FKI-5985

Aogacillins A and B, capable of overcoming arbekacin resistance in methicillin-resistant *Staphylococcus aureus* (MRSA), were isolated from a culture broth of *Simplicillium* sp. FKI-5985. Their structures were elucidated by NMR spectroscopic studies and ECD analyses. The aogacillins possessed a novel carbon skeleton, including a β -keto- γ -methyliden- δ -lactone ring connected to a 2-ethyl-6-methylcyclohexane ring by spiro conjugation.

Methicillin-resistant *Staphylococcus aureus* (MRSA) causes severe opportunistic infections, particularly in patients with compromised immune systems. MRSA has also been increasing significantly as a cause of serious community-acquired infection among healthy people. Although four antibiotics, arbekacin (ABK), vancomycin, teicoplanin, and linezolid, are approved as therapeutic agents for MRSA in Japan, microorganisms resistant to these compounds have increasingly been reported.

ABK has more potent bactericidal activity than other anti-MRSA agents, and it inhibits the growth of both Gram-positive and Gram-negative bacteria. ABK resistance

is believed to be caused by bacterial aminoglycosidemodifying enzymes, which phosphorylate and acetylate ABK to inactivate it.^{2,3} Consequently, any inhibitor of these enzymes can be used in combination with ABK to help maintain its effectiveness.

During our screening program to discover new anti-MRSA agents, we recently found four lactone compounds, biverlactones A–D,⁴ from the culture broth of *Penicillium* sp. FKI-4429, plus aranorosin from the culture broth of *Gymnascella aurantiaca*. All are effective ABK-resistance

[†] Tokyo University of Science.

¹ These authors contributed equally.

^{*}Kitasato Institute for Life Sciences, Kitasato University.

[§] School of Pharmacy, Kitasato University.

Wesleyan University.

⁽¹⁾ Diep, B. A.; Sensabaugh, G. F.; Somboonna, N. S.; Carleton, H. A.; Perdreau-Remington, F. J. Clin. Microbiol. 2004, 42, 2080–2084.

⁽²⁾ Tabata, M.; Shimizu, M.; Araake, M.; Ogawa, H. *Jpn. J. Antibiot.* **2003**, *56*, 36–43(in Japanese).

⁽³⁾ Ishino, K.; Ishikawa, J.; Ikeda, Y.; Hotta, K. J. Antibiot. 2004, 57, 679–686.

⁽⁴⁾ Iwatsuki, M.; Ishimori, T.; Yamamoto, T.; Takata, K.; Mori, M.; Nonaka, K.; Masuma, R.; Hayakawa, Y.; Hanaki, H.; Shiomi, K.; Omura, S. *Tetrahedron* **2011**, *67*, 6644–6648.

⁽⁵⁾ Suga, T.; Ishii, T.; Iwatsuki, M.; Yamamoto, T.; Nonaka, K.; Masuma, R.; Matsui, H.; Hanaki, H.; Ōmura, S.; Shiomi, K. *J. Antibiot.* **2012**, *65*, 527–529.

circumventors.⁵ Further research led to the discovery of two new compounds which could also overcome ABK resistance in MRSA, designated as aogacillins A (1) and B (2), obtained from culture broth of *Simplicillium* sp. FKI-5985 (Figure 1). In this paper, the isolation, structure elucidation, and biological activity of the two aogacillins are described.

A sample of the fungus *Simplicillium* sp. FKI-5985 was collected in Aogashima, Tokyo, Japan. Ethanol (3 L) was added to the culture broth at the end of fermentation. After filtration and removal of the ethanol *in vacuo*, the remaining aqueous solution was applied to an ODS silica gel column, and then eluted fractions containing 1 and 2 were further purified by reversed-phase HPLC to afford 1 (15.0 mg) and 2 (15.5 mg).

Aogacillin A (1) was obtained as yellow syrups. The molecular formula of 1 was elucidated by HR-FAB-MS to be $C_{15}H_{22}O_4$ (m/z 267.1589 [M+H]⁺ calcd 267.1597), requiring five degrees of unsaturation. The IR absorption at 1705, 1745, and 3420 cm⁻¹ suggested the presence of ketone carobonyl, ester carbonyl, and hydroxyl groups,

Figure 1. Structures of aogacillins A (1) and B (2).

respectively. The ¹H and ¹³C NMR spectral data of 1 are listed in Table 1. The ¹³C NMR, HSQC, and HMBC spectra indicated 15 carbons, which were classified into two carbonyl carbons, one sp^2 methylene carbon, one sp^2 quaternary carbon, two oxygenated sp^3 quaternary carbons, two sp^3 methine carbons, four sp^3 methylene carbons, and three methyl carbons, thus accounting for three degrees of unsaturation. Therefore, the remaining two degrees of unsaturation should be due to two ring structures. As shown by the bold lines for 1 in Figure 2, two partial structures were revealed by proton spin networks from H-7 ($\delta_{\rm H}$ 1.42) to H₃-15 ($\delta_{\rm H}$ 0.82) and from H₂-8 ($\delta_{\rm H}$ 1.38, 1.81) to H_3 -16 (δ_H 0.95) in the 1H – 1H COSY. On the basis of ¹H-¹³C HMBC experiments, the correlations from H₂-8 to C-14 ($\delta_{\rm C}$ 23.9) and from H₂-14 ($\delta_{\rm H}$ 0.92, 1.24) to C-8 ($\delta_{\rm C}$ 27.3) indicated the C-8 is linked to C-7. The correlations from H₃-16 ($\delta_{\rm H}$ 0.95), H₂-10 ($\delta_{\rm H}$ 1.50, 1.62), and H₂-14 to C-6 ($\delta_{\rm C}$ 91.6) in HMBC spectra revealed the presence of a 1-ethyl-3-methylcyclohexane moiety by the connection of C-7 and C-11 via C-6. The correlations from H₂-13 ($\delta_{\rm H}$ 5.77, 6.44) to C-4 ($\delta_{\rm C}$ 197.4), C-5 ($\delta_{\rm C}$ 145.5), and C-6 and from H₃-12 ($\delta_{\rm H}$ 1.54) to C-2 $(\delta_{\rm C}$ 174.9), C-3 $(\delta_{\rm C}$ 77.8), and C-4 established the alignment from C-2 to C-6 and attachment of C-13 to C-5. A δ-lactone ring formed by the connection between C-2 and C-6 via an oxygen atom was suggested by the characteristic ester carbonyl absorption at 1745 cm⁻¹ in the IR spectrum. The broad signal (δ_H 3.52 in CDCl₃) in ¹H NMR indicated the presence of a hydroxyl group. This was confirmed by the presence of a hydroxyl group at C-3 by high-field shifts of C-3 from δ_c 77.818 (in CD₃OD) to δ_c 77.782 (in CD₃OH) in the H/D exchange experiment. Thus, the structure of 1 was elucidated to be a novel compound

Table 1. NMR Data of Aogacillins A (1) and B (2) in CD₃OD

	Aogacillin A (1)			Aogacillin B (2)		
no.	$\delta_{ m H}$ (mult, J in Hz)	$\delta_{ m C}$	gHMBC	$\delta_{ m H} ({ m mult}, J { m in Hz})$	$\delta_{ m C}$	gHMBC
2	_	174.9		_	175.0	
3	_	77.8		_	77.9	
4	_	197.4		_	197.6	
5	_	145.5		_	145.6	
6	_	91.6		_	91.9	
7	1.42 (m)	54.6	8, 15	1.84 (m)	48.7	
8a	1.38 (m)	27.3	9, 14	1.36 (m)	26.4	9
8b	1.81 (m)			1.91 (m)		
9a	1.45 (m)	26.2	8	1.46 (m)	26.2	
9b	1.83 (m)			1.84 (m)		8
10a	1.50 (dddd, 14.0, 8.8, 8.8, 6.2)	30.2		1.52 (2H, m)	31.5	6, 11
10b	1.62 (dddd, 14.0, 6.4, 2.2, 2.2)		6, 9, 11			
11	2.12 (dqd, 8.8, 7.2, 5.4)	42.4	10, 16	1.68 (m)	48.1	6, 10, 16
12	1.54 (s)	27.2	2, 3, 4	1.55(s)	27.3	2, 3, 4
13a	5.77 (s)	125.5	3, 4, 5, 6	5.73 (s)	125.1	4, 5, 6
13b	6.44(s)		4, 5, 6	6.42 (s)		4, 5, 6
14a	0.92 (m)	23.9	7, 8, 15	1.14 (m)	24.7	7, 8, 15
14b	1.24 (m)		6, 7, 8, 15	1.66 (m)		6, 7, 8, 15
15	0.82 (t, 7.2)	12.3	7, 14	0.94 (t, 7.2)	12.3	7, 14
16	0.95 (d, 7.2)	17.2	6, 10, 11	0.71 (d, 7.2)	16.1	6, 10, 11

Org. Lett., Vol. 15, No. 18, 2013

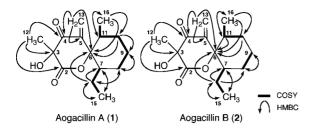


Figure 2. COSY and HMBC correlations of aogacillins A (1) and B (2).

including a β -keto- γ -methyliden- δ -lactone ring connected to a 2-ethyl-6-methylcyclohexane ring by spiro conjugation. Only a few natural product compounds having δ -lactone with a spiro-conjugated cyclohexane have been reported, but all of their cyclohexanes are fused with the other ring, such as vitexifolin E from the fruit of *Vitex rotundifolia*. The relative configuration of 1 was elucidated to be $3S^*, 6R^*, 7S^*, 11R^*$ by ROESY and differential NOE correlations of H-13a/H-7, H-13a/H-11, H-7/H-11, and H₃-16/H₃-12 (Figure 3a).

The molecular formula of 2 was elucidated by ESI-MS to be $C_{15}H_{22}O_4$ (m/z 289.1428 [M+Na]⁺ calcd 289.1416), the same mass units as 1. The ¹H and ¹³C NMR spectral data of 2 are listed in Table 1. The analysis of 1D and 2D NMR revealed 2 has the same planar structure to 1 as shown in Figure 2. Although the ¹H and ¹³C NMR spectra of 2 were very similar to those of 1 (Table 1), both ¹H and ¹³C chemical shifts of CH₂-14 and CH₃-15 shifted highfield (CH₂-14: $\Delta\delta_{H}\!=\,-0.22$ and -0.42 ppm, $\Delta\delta_{C}=-0.8$ ppm/CH₃-15: $\Delta \delta_{\rm H} = -0.12$ ppm) in 1 (Figure 3c). Conversely, those of CH₃-16 largely shifted low-field ($\Delta \delta_{\rm H} =$ +0.24 ppm, $\Delta\delta_{\rm C}=+1.1$ ppm) in 1. This observation is explainable by a magnetic anisotropic effect induced by the ester carbonyl group at C-2, on which the ethyl group (C-14–C-15) in **1** and the methyl group (C-16) in **2** are located. On the other hand, steric compression between H-13a and H-11 in 1 and between H-13a and H-7 in 2 caused opposite shifts of ¹H and ¹³C chemical shifts at C-11 $(\Delta \delta_{\rm H} = +0.44 \, {\rm ppm} \, {\rm and} \, \Delta \delta_{\rm C} = -6.7 \, {\rm ppm}) \, {\rm and} \, {\rm C}\text{-}7 \, (\Delta \delta_{\rm H} =$ -0.42 ppm and $\Delta\delta_{\rm C}$ = +5.9 ppm). This difference of chemical shifts and the correlations of ROESY and differential NOE of H-13a/H-7, H-13a/H-11, H₃-12/H₂-14, and H_3 -12/ H_3 -15 revealed the relative configuration of 2 to be $3R^*,6R^*,7S^*,11R^*$, and thus 2 is a C-3 epimer of 1 (Figure 1).

The absolute stereochemistries of 1 and 2 were deduced by electronic circular dichroism (ECD) spectra^{7,8} in comparison to their calculated spectra⁹ (Figure 4). The

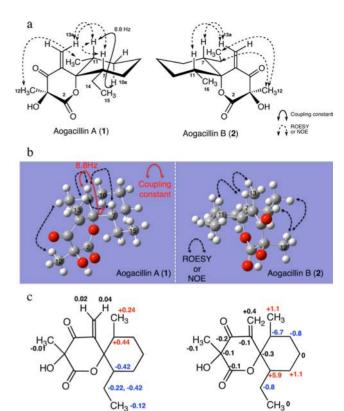


Figure 3. Elucidation of the relative stereochemistry of aogacillins A (1) and B (2); (a) key ROESY and differential NOE correlations, (b) key ROESY and differential NOE correlations with models of caliculated conformers, (c) difference of chemical shifts between 1 and 2.

 $\Delta \delta_{C} = \delta_{C} (1) - \delta_{C} (2)$

 $\Delta \delta_{H} = \delta_{H} (1) - \delta_{H} (2)$

conformational analyses of 1 and 2 started from MM2 force field structures further optimized by semiempirical PM3 calculations and then further refined by density functional theory (DFT) at the B3LYP/6-311G level of theory which yielded additional relevant conformers. Structures of resulting calculated conformers¹⁰ were supported by our NOE experiments (Figure 3b). The ECD spectra of 1 and 2 were simulated using time-dependent DFT (TD-DFT) at the same level of theory on the relevant conformers.

As a result of the comparison between the experimental ECD spectrum and calculated ECD spectrum, the absolute stereochemistry of $\bf 1$ was elucidated to be 3S,6R,7S,11R. In the same manner, that of $\bf 2$ was elucidated to be 3R,6R,7S,11R.

A plausible biosynthetic pathway for 1 and 2 is illustrated in Scheme 1. One acetate and five malonate units might be required to assemble the polyketide skeleton for 1 and 2. The linear polyketide precursor further undergoes a series of reactions, including methylation by S-methyladenosylmethionine (SAM), aldol cyclization to form a spiro skeleton,

Org. Lett., Vol. 15, No. 18, 2013

⁽⁶⁾ Ono, M.; Yanaka, T.; Yamamoto, M.; Ito, Y.; Nohara, T. J. Nat. Prod. 2002, 65, 537–541.

⁽⁷⁾ Stonard, R. J.; Trainor, D. A.; Nakatani, M.; Nakanishi, K. J. Am. Chem. Soc. 1983, 105, 130–131.

⁽⁸⁾ Li, X.-C.; Ferreira, D.; Ding, Y. Curr. Org. Chem. 2010, 14, 1678–1697.

⁽⁹⁾ Recent review: Nugroho, A. E.; Morita, H. J. Nat. Med. **2013**10.1007/s11418-013-0768-x.

⁽¹⁰⁾ Calculated conformers for compounds 1 and 2 are shown in the SI, together with their atomic coordinates and energies.

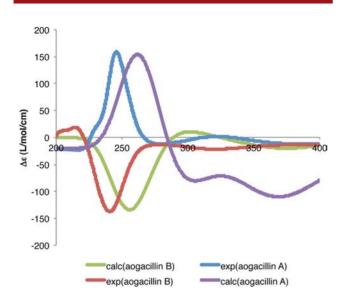


Figure 4. Experimental CD spectra of 1 and 2 overlaid with calculated spectra.

reduction, and then dehydrogenation. As a last step, oxidation at C-3 might produce two epimers at C-3 of the aogacillins.

We investigated the effect of aogacillins on the activity of ABK against ABK-resistant MRSA by the liquid microdilution method. The effect of aogacillins A (1) and B (2) were tested at first on the growth of MRSA, resulting in equivalent MIC values of 2.0 μ g/mL. Therefore, the concentration of aogacillins for combination with ABK was set at 0.5 μ g/mL (one-fourth of each MIC value, with no effect on growth of the MRSA) allowing us to investigate any effect on the activity of ABK against ABK-resistant MRSA.

Aogacillins A (1) and B (2) markedly reduced the MIC value of ABK against ABK-resistant MRSA, from 256 to 8 μ g/mL, which was similar to the MIC value of ABK against ABK-sensitive MRSA (1 μ g/mL). Moreover, we investigated the circumvention effect of aogacillins against ABK-resistant MRSA using various antibiotics by the liquid microdilution method (see Supporting Information

Scheme 1. Plausible Biosynthetic Pathways for Aogacillins A (1) and B (2)

(SI)). The concentrations of aogacillins used were $0.5\,\mu g/mL$ for each antibiotic. The aogacillins showed no effect, or only slight potentiation, with respect to the MIC values of aminoglycosides tested (kanamycin, dibekacin, and streptomycin) and other antibiotics (ampicillin, ciprofloxacin, and vancomycin) against the MRSA (1–4-fold). These data suggested that the aogacillins could be a specific circumventor of ABK activity for ABK-resistant MRSA. Investigation of the mode of action of the aogacillins is now underway.

Acknowledgment. This study was supported, in part, by funds from the Quality Assurance Framework of Higher Education from the Ministry of Education, Culture, Sports, Science and Technology (MEXT) in Japan and by a Grant-in-Aid for Scientific Research (C, 21580129) from the Japan Society for the Promotion of Science. We are grateful to Ms. Noriko Sato and Dr. Kenichiro Nagai (School of Pharmacy, Kitasato University) for measurements of MS and NMR.

Supporting Information Available. Experimental procedures; UV, IR, ¹H NMR, ¹³C NMR, gCOSY, gHSQC, gHMBC, and ROESY spectra of aogacillins A (1) and B (2). This material is available free of charge via the Internet at http://pubs.acs.org.

The authors declare no competing financial interest.

Org. Lett., Vol. 15, No. 18, 2013